Mechanical Engineering and Allied branches(Chemistry group)

CourseTitle:	Applied Chemistry for Mechanical Engineering stream		
CourseCode:	BCHEM202/202	CIEMarks	50
Course Type(Theory/Practical/Integrated)	Integrated	SEEMarks	50
		Total Marks	100
TeachingHours/Week(L:T:P:S)[1]	2:2:2:0	Exam Hours	03
TotalHoursofPedagogy	40hoursTheory+10to12Labslots	Credits	04

Courseobjectives

- Toenablestudentstoacquireknowledgeonprinciplesofchemistryforengineeringapplications.
- Todevelopanintuitiveunderstandingofchemistrybyemphasizingtherelatedbranchesofengineering.
- Toprovidestudentswithasolidfoundationinanalyticalreasoningrequiredtosolvesocietal problems.

Teaching-LearningProcess

Thesearesamplestrategies,whichteachercanusetoacceleratetheattainmentofthevariouscourseoutcomesandmakeTeaching–Learningmoreeffective

- Tutorial&remedialclassesforneedystudents(notregularT/R)
- ConductingMakeupclasses/Bridgecoursesforneedystudents
- Demonstrationofconceptseitherbybuildingmodelsorbyindustryvisit
- Experimentsinlaboratoriesshallbeexecutedinblendedmode(conventionalornon–conventionalmethods)
- UseofICT–Onlinevideos,onlinecourses
- Useofonlineplatformsforassignments/Notes/Quizzes(Ex.Googleclassroom)

Module-1:Energy;Source,ConversionandStorage(8hr)

Fuels: Introduction,calorificvalue,determinationofcalorificvalueusingbombcalorimeter, numericalproblemsonGCVandNCV.

Greenfuels: Introduction,poweralcohol,synthesisandapplicationsofbiodiesel.

High energy fuels: Production of hydrogen by electrolysis of water and its advantages.**Energy devices:** Introduction, construction, working, and applications of Photovoltaic cells,Li-ionbatteryandmethanol-oxygen fuelcell.

Self-learning: Plasticrecyclingtofuelsandits monomersorotherusefulproducts.

Module-2:CorrosionScienceandEngineering(8hr)

Corrosion: Introduction,electrochemicaltheoryofcorrosion,typesofcorrosion-differential metal, differential aeration (waterline and pitting),stress corrosion (causticembrittlement).

Corrosioncontrol: Metalcoating-galvanization,surfaceconversioncoating-anodizationand cathodic protection-sacrificial anode method. Corrosion testing by weight loss method.Corrosionpenetrationrate (CPR)-numericalproblems.

Metalfinishing: Introduction,technologicalimportance.Electroplating:Introduction,

1.NOTE:Whereverthecontacthoursarenotsufficient,tutorialhourscanbeconvertedto theoryhours.

Electroplatingofchromium(hardanddecorative).Electrolessplating:Introduction,electrolessplatingofnickel.

Self-learning: Factorsaffectingtherateofcorrosion,**f**actorsinfluencingthenatureand qualityofelectrodeposit(Currentdensity, concentrationofmetalion,pHandtemperature)**.**

Module-3:MacromoleculesforEngineeringApplications(8hr)

Polymers:Introduction,methodsofpolymerization(CondensationandFreeradical),molecular weight; number average andweight average,numericalproblems.Synthesis,propertiesandindustrialapplicationsofpolyvinylchloride(PVC)andpolystyrene.

Fibers: Introduction,synthesis,propertiesandindustrialapplicationsofKevlarandPolyester.

Plastics: Introduction,synthesis,propertiesandindustrialapplicationsofpoly(methylmethacrylate)(PMMA)andTeflon.

Composites: Introduction,propertiesandindustrialapplicationsofcarbon-basedreinforced composites (graphene/carbon nano-tubes as fillers) and metal matrix polymercomposites.

Lubricants:Introduction,classification,propertiesandapplicationsoflubricants.

Self-learning: Biodegradable polymer: Introduction, synthesis, properties and applicationsofpolylacticacid(PLA).

Module-4:PhaseRuleandAnalyticalTechniques(8hr)

Phase rule: Introduction, Definition of terms: phase, components, degree of freedom, phaseruleequation.Phase diagram:Twocomponent-lead-silversystem.

Analytical techniques: Introduction, principle, instrumentation of potentiometric sensors;its application in the estimation of iron, Optical sensors (colorimetry); its application in theestimationofthecopper,pH-sensor(Glasselectrode);itsapplicationinthedeterminationofpH ofbeverages.

Self-learning: Determinationofviscosityofbiofuelanditscorrelationwithtemperature.

Module-5:MaterialsforEngineeringApplications(8hr)

Alloys:Introduction,classification,composition,propertiesandapplicationsofStainlessSteel,BrassandAlnico.

Ceramics:Introduction,classificationbasedonchemicalcomposition,propertiesandapplications ofperovskites($CaTiO_3$).

Nanochemistry: Introduction,size-dependentpropertiesofnanomaterial(surfacearea,catalytical and thermal), synthesis of nanoparticles by sol-gel, and co-precipitation method.**Nanomaterials:** Introduction, properties and engineering applications of carbon nanotubesandgraphene.

Self-learning:Abrasives: Introduction,classification, propertiesandapplicationsof silicon carbide(carborundum).

PRACTICALMODULE

A–Demonstration(anytwo)offline/virtual:

A1.Synthesisofpolyurethane

A2.PreparationofureaformaldehyderesinA

3. Synthesis of iron oxide nanoparticlesA4.Determinationofacidvalue ofbiofuel

B–Exercise(compulsorilyany4tobeconducted):

B1.Conductometricestimationofacidmixture
B2.PotentiometricestimationofFASusing$K_2Cr_2O_7$
B3.DeterminationofpKaofvinegarusingpHsensor(Glasselectrode)
B4.Determinationofrateofcorrosionofmildsteelbyweightlossmethod B5. Estimationoftotalhardnessofwaterby EDTA method

C–StructuredEnquiry (compulsorilyany4tobeconducted):

C1. Estimation of Copper present in electroplating effluent by optical sensor (colorimetry) C2.DeterminationofViscositycoefficientoflubricant(Ostwald'sviscometer)
C3. Estimation of iron in TMT bar by diphenyl amine/external indicator method C4.EstimationofSodiumpresentinsoil/effluentsampleusingflamephotometry
C5.DeterminationofChemicalOxygenDemand(COD)ofindustrialwastewatersample

D–OpenEndedExperiments(anytwo):

D1.Estimationofpercentageofironinsteel
D2.ElectroplatingofdesiredmetalonsubstrateD3.Synthesisofbiodiesel
D4.SynthesisofAluminiumOxidenanoparticle

Courseoutcome(CourseSkillSet): Attheendofthecourse,thestudentwillbeableto:

CO1.	Identify the terms processes involved in scientific and engineering andapplications
CO2.	Explainthephenomenaofchemistrytodescribethemethodsofengineering processes
CO3.	Solvetheproblemsin chemistrythatarepertinentinengineeringapplications
CO4.	Applythebasicconceptsofchemistrytoexplainthechemicalpropertiesandprocesses
CO5.	Analyze properties processes associated with chemical substances in andmultidisciplinarysituations

AssessmentDetails(bothCIEandSEE)

The weightage of Continuous Internal Evaluation (CIE) is 50% and for Semester End Exam (SEE) is50%. The minimum passing mark for the CIE is 40% of the maximum marks (20 marks out of 50).The minimum passing mark for the SEE is 35% of the maximum marks (18 marks out of 50). Astudent shall be deemed to have satisfied the academic requirements and earned the credits allottedtoeach subject/course ifthestudentsecuresnotlessthan35%(18Marksoutof50)in thesemester-end examination (SEE), and a minimum of 40% (40 marks out of 100) in the sum total oftheCIE(ContinuousInternalEvaluation)andSEE(SemesterEndExamination)takentogether.

ContinuousInternalEvaluation(CIE):

The CIE marks for the theory component of the IC shall be **30 marks** and for the laboratory component **20 Marks**.

CIE for the theory component of the IC

- Three Tests each of 20 Marks**;** after the completion of the syllabus of 35-40%, 65-70%, and 90-100% respectively.
- Two Assignments/two quizzes/ seminars/one field survey and report presentation/one-course project totalling 20 marks.

Total Marks scored (test + assignments) out of 80 shall be scaled down to **30 marks**

CIE for the practical component of the IC

- On completion of every experiment/program in the laboratory, the students shall be evaluated and marks shall be awarded on the same day. The **15 marks** are for conducting the experiment and preparation of the laboratory record, the other **05 marks shall be for the test** conducted

at the end of the semester.

- The CIE marks awarded in the case of the Practical component shall be based on the continuous evaluation of the laboratory report. Each experiment report can be evaluated for 10 marks. Marks of all experiments' write-ups are added and scaled down to 15 marks.
- The laboratory test **(duration 03 hours)** at the end of the 15th week of the semester /after completion of all the experiments (whichever is early) shall be conducted for 50 marks and scaled down to **05 marks.**

Scaled-down marks of write-up evaluations and tests added will be CIE marks for the laboratory component of IC/IPCC for **20 marks**.

- The minimum marks to be secured in CIE to appear for SEE shall be 12 (40% of maximum marks) in the theory component and 08 (40% of maximum marks) in the practical component. The laboratory component of the IC/IPCC shall be for CIE only. However, in SEE, the questions from the laboratory component shall be included**.** The maximum of 05 questions is to be set from the practical component of IC/IPCC, the total marks of all questions should not be more than 25 marks.

The theory component of the IC shall be for both CIE and SEE.

Semester End Examination(SEE):

Theory SEE will be conducted by University as per the scheduled timetable, with common question papers for the subject (**duration 03 hours)**

- The question paper shall be set for 100 marks. The medium of the question paper shall be English/Kannada). The duration of SEE is 03 hours.
- The question paper will have 10 questions. Two questions per module. Each question is set for 20 marks. The students have to answer 5 full questions, selecting one full question from each module. The student has to answer for 100 marks and **marks scored out of 100 shall be proportionally reduced to 50 marks**.

There will be 2 questions from each module. Each of the two questions under a module (with a maximum of 3 sub-questions), **should have a mix of topics** under that module.

SuggestedLearningResources:

Books(TitleoftheBook/Nameoftheauthor/Nameofthepublisher/EditionandYear)

1. WileyEngineeringChemistry,WileyIndiaPvt.Ltd.NewDelhi,2013-2nd Edition.
2. EngineeringChemistry,Satyaprakash&ManishaAgrawal,KhannaBookPublishing,Delhi
3. ATextBookofEngg.Chemistry,ShashiChawla,DhanpatRai&Co.(P)Ltd.
4. EssentialsofPhysicalChemistry,Bahl&Tuli,S.ChandPublishing
5. AppliedChemistry,SunitaRattan,Kataria5.EngineeringChemistry,Baskar,Wiley
6. EngineeringChemistry–I,D.GrourKrishana,VikasPublishing
7. ATextbookofEngineeringChemistry,SSDara&Dr.SSUmare,SChand&CompanyLtd., 12th Edition, 2011.
8. ATextBookofEngineeringChemistry,R.V.GadagandNityanandaShetty,I.K.InternationalPublishinghouse. 2nd Edition,2016.
9. TextBookofPolymerScience,F.W.Billmeyer,JohnWiley&Sons,4th Edition,1999.
10. NanotechnologyAChemicalApproachtoNanomaterials,G.A.Ozin&A.C.Arsenault,RSCPublishing,2005.
11. CorrosionEngineering,M.G.Fontana,N.D.Greene,McGrawHillPublications,NewYork,3rd Edition,1996.
12. Linden'sHandbookofBatteries,KirbyW.Beard,FifthEdition,McGrawHill,2019.
13. OLEDDisplayFundamentalsandApplications,TakatoshiTsujimura,Wiley–Blackwell,2012
14. Supercapacitors:Materials,Systems,andApplications,MaxLu,FrancoisBeguin,ElzbietaFrackowiak, Wiley-VCH;1stedition,2013.
15. "HandbookonElectroplatingwithManufactureofElectrochemicals",ASIAPACIFICBUSINESSPRESS Inc., 2017. Dr.H. Panda,

16. ExpandingtheVisionofSensorMaterials.NationalResearchCouncil1995,Washington,DC:TheNationalAcademies Press. doi:10.17226/4782.
17. EngineeringChemistry,EditedbyDr.MaheshBandDr.RoopashreeB,SunstarPublisher,Bengaluru,ISBN978-93-85155-70-3, 2022
18. HighPerformanceMetallicMaterialsforCostSensitiveApplications,F.H.Froes,etal.JohnWiley&Sons, 2010
19. InstrumentalMethodsofAnalysis,Dr.K.R.Mahadik and Dr.L.Sathiyanarayanan,NiraliPrakashan,2020
20. PrinciplesofInstrumentalAnalysis,DouglasA.Skoog,F.JamesHoller,StanleyR.CrouchSeventhEdition, CengageLearning, 2020
21. PolymerScience,VRGowariker,NVViswanathan,Jayadev,Sreedhar,NewageInt.Publishers,4thEdition, 2021
22. EngineeringChemistry,PCJain&MonicaJain,DhanpatRaiPublication,2015-16thEdition.
23. Nanostructuredmaterialsandnanotechnology,HariSingh,Nalwa,academicpress,1stEdition,2002.
24. NanotechnologyPrinciplesandPractices,SulabhaKKulkarni,CapitalPublishingCompany,3rdEdition 2014
25. Principlesofnanotechnology,Phanikumar,Scitechpublications,2ndEdition,2010.
26. ChemistryforEngineeringStudents,B.S.JaiPrakash,R.Venugopal,Sivakumaraiah&PushpaIyengar.,SubashPublications,5thEdition, 2014
27. "EngineeringChemistry",O.G.Palanna,TataMcGrawHillEducationPvt.Ltd.NewDelhi,FourthReprint, 2015.
ChemistryofEngineeringmaterials,MaliniS,KSAnanthaRaju,CBSpublishersPvtLtd.,
28. LaboratoryManualEngg.Chemistry,AnupmaRajput,DhanpatRai&Co.

WeblinksandVideoLectures(e-Resources):

- http://libgen.rs/
- https://nptel.ac.in/downloads/122101001/
- https://nptel.ac.in/courses/104/103/104103019/
- https://ndl.iitkgp.ac.in/
- https://www.youtube.com/watch?v=faESCxAWR9k
- https://www.youtube.com/watch?v=TBqXMWaxZYM&list=PLyhmwFtznRhuz8L1bb3X-9IbHrDMiHWWb

28. LaboratoryManualEngg.Chemistry,AnupmaRajput,DhanpatRai&Co.
28. Laboratory Manual Engg. Chemistry, Anupma Rajput, DhanpatRai&Co.

- https://www.youtube.com/watch?v=X9GHBdyYcyo
- https://www.youtube.com/watch?v=1xWBPZnEJk8
- https://www.youtube.com/watch?v=wRAo-M8xBHM

ActivityBasedLearning(SuggestedActivitiesinClass)/PracticalBasedlearning

- https://www.vlab.co.in/broad-area-chemical-sciences
- https://demonstrations.wolfram.com/topics.php
- https://interestingengineering.com/science

16-2-2023

COsandPOsMapping(Individualteacherhastofillup)												
	PO											
	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
CO1	3	1	1				1					
CO2	3	1	1				1					
CO3	3	1	1				1					
CO4	3	1	1				1					
CO5	3	1	1				1					

16-2-2023